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Name:

Date:

Materials: Metals and Non-Metals

Q1. State some of the chemical properties of metals.

Ans.



Q2. Some properties are listed in the following Table. Distinguish between metals and non-metals on the basis of these properties.

Ans.

Properties	Metals	Non-metals
1. Appearance		
2. Hardness		
3. Malleability		
4. Ductility		
5. Heat Conduction		
6. Conduction of		
Electricity		



Materials: Metals and Non-Metals

Q1. State some of the chemical properties of metals.

Ans. <u>Chemical properties of metals</u>

i. On burning, metals react with oxygen to produce metal oxides which are basic in nature.

ii. Some metals react with water to produce metal hydroxides and hydrogen gas.

- iii. Metals react with acids and produce metal salts and hydrogen gas.
- iv. Some metals react with bases to produce hydrogen gas.
- v. More reactive metals displace less reactive metals from their compounds in aqueous solutions.
- Q2. Some properties are listed in the following Table. Distinguish between metals and non-metals on the basis of these properties.

Ans.

Properties	Metals	Non-metals
1. Appearance	Lustrous	Dull
2. Hardness	Hard	Soft
3. Malleability	Can be beaten into thin	Cannot be beaten into
	sheets.	thin sheets.
4. Ductility	Can be drawn into wires.	Cannot be drawn into
		wires.
5.Heat Conduction	Good conductor of heat.	Bad conductor of heat.
6. Conduction of	Good conductors of	Bad conductors of
Electricity	Electricity.	Electricity.